Part 1
Metabolic Pathways

Roche Biochemical Pathways
4th Edition, Part 1 – Editor: Gerhard Michal

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Roche

General Remarks for Part 1 and 2

1) Enzymes are in blue, EC-Numbers (Recommended Names) in the "Enzyme Nomenclature 1992" and "Supplement 1 (1993 – 94)" with its supplements in Eur. J. Biochem. 167, 1 (1987) and 179, 489 (1989) are used. Enzymes not listed there are printed in parentheses.
2) Coenzymes are shown in red, other reaction participants in black.
3) Color of the arrows shows type of organism in which the reaction was observed or is typical:
   - green: plants
   - violet: animals
   - blue: microorganisms (bacteria + fungi)
   - Dashed arrows are used for catabolic pathways, continuous lines for anabolic and unspecific reactions. Arrows pointing in both directions indicate reversible reactions, double arrows point in the preferred direction.
4) Regulatory effects are shown in orange. Metal ions and similar activators are located above the reaction arrow, inhibitors below. Only the most typical, or the most important mechanisms have a continuous orange arrow coming from the effector site with a "blue" regulation (e.g. induction or repression of the enzyme synthesis) a dashed orange arrow. Where the mechanism of regulation is complex, several arrows may coincide, that the only one of multiple arrows is significant for one isoenzyme. Regulatory effects of hormones are not shown in this overview.
5) Compounds enclosed in black boxes; sharp edged boxes occur in several places in the pathways, rounded boxes only at one place. Cofactors are not boxed.
6) Boxes with hatched corners are sometimes used in part 2 to indicate enzyme clusters.
7) Compounds in brackets [ ] are unstable.
8) Compound names in orange boxes indicate excretory or end products of metabolism.
9) Structural formulas are given for the most common isomer at pH 7 (physiological pH). Structural formulas, unless they fit the text and / or have a special meaning, are drawn against different grades of ionization occur simultaneously. Therefore, participation of H+ in the reaction equations is frequently not balanced.
10) Organic phosphates: P in "inorganic phosphate" is not shown, phosphoric acids is drawn.
11) All amino acids can be attached to their respective RNAs by the appropriate ligases. This is not shown in the pathways unless it is relevant.
12) Cellular surface structures of the eukaryotic cell, like cell walls, cell membranes, cell organelles etc. have approximately the same thickness. For graphical reasons, however, they are drawn with different thicknesses.
13) RNA molecules are drawn with solid lines, DNA with double lines.
14) Molecular weights of proteins are given in kDa (kilodalton) as a subscript number designating them e.g. ^{14}A^{14}^{kDa}, indicating the molecular weight (in kD) and the origin (A = cellular, V = viral).

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Abbreviations for Part 1

ACP	acceptor (in redox reactions)
ACTH	adrenocorticotropic hormone
AdoMet	adenosylmethionine, S-adenosylmethionine (SAM)
ALA	aminolevulinic acid
CAMP	cyclic AMP
CDP	cytidine diphosphate
CMP	cytidine monophosphate
DHF	dihydrofolate
DNA	deoxyribonucleic acid
dUMP	deoxyuridine monophosphate
FAD	flavin adenine dinucleotide
FADH$_2$	reduced flavin adenine dinucleotide
FMN	flavin mononucleotide
FMNH$_2$	reduced flavin mononucleotide
GDP	guanosine diphosphate
GMP	guanosine monophosphate
GSH	glutathione, reduced
GSSG	glutathione, oxidized
GTP	guanosine triphosphate
IMP	inosine monophosphate
mRNA	messenger RNA
NAD$^+$	nicotinamide adenine dinucleotide
NADH	reduced nicotinamide adenine dinucleotide
NADP$^+$	nicotinamide adenine dinucleotide phosphate
NADPH	reduced nicotinamide adenine dinucleotide phosphate
PAPS	phosphoadenosine phosphosulfate
PEP	phosphoenolpyruvate
PLP	pyridoxal phosphate
PRPP	phosphoribosyl pyrophosphate
RNA	ribonucleic acid
rRNA, tRNA, mRNA	ribosomal-, transfer-, messenger RNA
R-S-S-R	oxidized thiol
R-SH	thiol
SAM	S-adenosylmethionine
TDP	thiamine diphosphate
TPP	thiamine pyrophosphate
UDP	uridine diphosphate
UMP	uridine monophosphate
UTP	uridine triphosphate

Notes for Part 1

1) Gluconolactone is also D-glucono-δ- also split by this enzyme.
2) Glucose-6-phosphate dehydrogenase is also found in bacteria. The latter compound in turn is oxidized by gluconolactone dehydrogenase or L-gluconolactone dehydrogenase.
3) RNA directed RNA polymerase (replicase) is active in RNA virus infected cells.
4) Choline oxidase also oxidizes betaine aldehyde to betaine forming H$_2$O$_2$ after H$_2$O$_2$ is split by catalase. Choline dehydrogenase is found in bacteria and mammalian liver.
5) Betaine homocysteine S-methyltransferase is found in liver and kidney.
6) Glycine cleavage system is also found in bacteria (requiring Cu) and plants (requiring Mn).
7) In mitochondria the fatty acid is removed from AcCP by ribaC (ACP) hydrolase and is subsequently converted to acyl-CoA.
8) Fatty acid synthase is a complex of seven different enzymes, malonyl-, acetyl-, palmitoyl-transferase, 3-ketoacyl-reductase, -dehydrase, enoyl-reductase, and ACP. Only the intermediates in the glucosinate conversion by indicated enzymes in *E.coli* are shown here and in Compendium Biochemistry, 4th Edition, Part 2.
9) Glucosinolates.
10) In some microorganisms,cobalamine synthesis takes place via Co-ated(-) hydrogenobyrinic acid.
11) The enzyme ferrochelatase for formation of heme from protoporphyrin and Fe^{++} is also found in bacteria. Fe^{+++} is reduced to Fe^{++} by NADPH + H$^+$ and ferredoxin.
12) Bilirubin is also known from the central 3rd group is written at the bottom-the marginal 3rd group at the top. In animals and in *E.coli*, the enzyme, biliverdin reductase, is NADPH-dependent.
13) Examples are D2H-homocystine transhydrogenase and protein disulfide reductase.
14) Ferredoxin is a general name for proteins with Fe-S centers. Their role as electron donors for reduction of nucleotide triphosphates takes place in Lactobacillus fermentation, *E.coli*, Klebsiella and Rhodobium reductases may be replaced by glutaredoxin (plus GSH) and GSH reductase.
15) Ribonucleotide reductase is also found in Lactobacillus leichmanii.
16) Thymidylate synthase is also found in bacteria. It differs from the animal enzyme in cofactors and purine groups, forming 5-methylene. Other splitting types are also known.
17) In many species the biosynthesis of purines, pyrimidines,amino acids,fatty acids etc. is regulated by feedback inhibition of the first enzyme of the pathway.
18) NADP is used by animal and yeast enzymes,NAD by the bacterial enzyme.
19) Glucose-6-phosphate and -gluconate are also found in bacteria, but not in mammals.
20) Formation of α- and β-ureidene occurs via ureidene-α-carotene may also be formed from β-carotene. Lycopene is also found in bacteria and cyanobacteria (blue-green algae).
21) Biosynthesis of cholesterol in mammalian systems is controlled by the regulation of metabolites or via S-adenosylmethionine and further steps similar to Pseudomonas.
22) Ubiquinone is also known as coenzyme Q.
23) The mammalian somatotrophic pathway is mostly used by bacteria (e.g.Pseudomonas).
24) Cytochrome P-450 is also found in bacteria and yeast.
25) β-adrenergic(1,3-methylthiopropionic is formed from S-adenosylmethionine by transmethylation. Cobalamine deoxyribosuside is involved in spermidine and spermine synthesis.
26) Enzymes of this function are known (oxidizing) - 5-Hydroxyisocitrate - 3-Hydroxy-3-methylglutaryl-CoA - 3-Hydroxy-3-methylglutaryl-CoA - 3-Hydroxy-3-methylglutaryl-CoA
27) The 2-oxo-propionic acid formed in the reaction is also known as pyruvic acid.
28) Lactate dehydrogenase is activated by phosphorylation catalyzed by protein kinase. Pyruvate dehydrogenase is activated by dephosphorylation by a specific phosphatase.
29) Pyruvate dehydrogenase is also found in bacteria. It differs from the animal enzyme in cofactors and purine groups, forming 5-methylene. Other splitting types are also known.
30) In Pseudomonas, the reaction proceeds via S-aminovaleronitrile,5-aminovalerate and 5-aminovaleraldehyde.
31) E. coli, the reaction proceeds via 2-hydroxy-3-keto-valerate and further steps similar to Pseudomonas.
32) Ethanolamine is also found in bacteria.
33) Examples are D2H-homocystine transhydrogenase and protein disulfide reductase.
34) Ferredoxin is a general name for proteins with Fe-S centers. Their role as electron donors for reduction of nucleotide triphosphates takes place in Lactobacillus fermentation, *E.coli*, Klebsiella and Rhodobium reductases may be replaced by glutaredoxin (plus GSH) and GSH reductase.
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